



Enhanced performance of planar perovskite solar cells using $\text{TiO}_2/\text{SnO}_2$ and TiO_2/WO_3 bilayer structures: Roles of the interfacial layers

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ABSTRACT

In planar perovskite solar cells (PSCs), engineering the extraction and recombination of electron–hole pairs by modification of the electron transport layer (ETL)/perovskite interface is very vital for obtaining high performance. The main idea here is to improve properties of the TiO_2 /perovskite interface by inserting an ultra-thin layer (UTL) of WO_3 or SnO_2 with the thickness of less than 10 nm by RF magnetron sputtering method. The structural and electrical characteristics of the samples were tested by XRD, AFM, FE-SEM, Mott-Schottky analysis, UV–Vis spectroscopy, J–V characterization and electrochemical impedance spectroscopy (EIS). It was found that the bilayer structured ETLs exhibit advantages of high electron extraction and low interfacial recombination together mainly based on a more effective energy level alignment. The best performance here belongs to the cell with $\text{TiO}_2/\text{SnO}_2$ bilayer by which an improved efficiency exceeding 12% was achieved for the planar PSCs. The role of SnO_2 and WO_3 UTLs was also modeled using SCAPS-1D software. This simulation supported the experimental results and was used to interpret the photovoltaic behaviors of the fabricated devices based on defect characteristics. The present work proposes a simple and promising method to make photovoltaic devices with desirable charge transport and recombination properties.

1. Introduction

Utilization of solar energy and its conversion into electricity, as the most convenient for use in modern life, is one of the major challenges of the twenty-first century. Demands for improving performance and cost-effectiveness of harnessing the solar energy has made it indispensable to get progress in photovoltaics. Perovskite solar cells (PSCs), composed of the organic-metal-halide materials, have shown enormous potential as cheaper alternatives to existing conventional solar cells. They have made impressive progress in maximum power conversion efficiencies (PCEs) from 3.8% to 25.2% in just one decade (NREL, 2019).

The electron transport layer (ETL) plays a significant role in achieving high efficiency in PSCs (Chen et al., 2018). The characteristics of the ETL, including its energy level alignment, trap states, charge mobility and morphology-dependent parameters, material and its

related interface properties are very vital in PSCs performance (Yang et al., 2016b). Mesoporous ETLs (~150 nm) are normally used in the stable and highly efficient PSCs, but they offer a relatively complicated device fabrication and their large scale manufacturing by spin-coating is still challenging. Therefore, the planar device structure has attracted significant attention for its feasible and roll-to-roll manufacturing capability. However, the planar PSCs exhibit significantly lower PCE and higher hysteresis compared to the mesoporous-type PSCs, owing to defects in the ETL and direct contacts between the perovskite and FTO layers (Ding et al., 2018).

Many modifications have been applied on ETLs to improve the efficiency of planar PSCs such as doping, finding the best material, controlling physical properties, composition and interface engineering (El Haimour et al., 2020; Mohammadian-Sarcheshmeh and Mazloum-Ardakani, 2018; Yang et al., 2016b). The interface engineering

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includes adding the self-assembled monolayers (Yang et al., 2016b), ETL surface treatments (Ke et al., 2015; Li et al., 2015) and introducing the organic molecules (Yang et al., 2016a). Inserting a wide band gap metal oxide semiconductor (such as ZnO, SrO and MgO) ultra-thin layer (UTL) at ETL's interfaces has been also reported as a capable approach in improving electron extraction and suppressing charge recombination (Mohammadian-Sarcheshmeh and Mazloum-Ardakani, 2018; Yang et al., 2016b). There are examples of successfully building an effective band alignment (Kulkarni et al., 2016; Lu et al., 2017b; Xu et al., 2015), providing a base for better formation of the perovskite layer (Kogo et al., 2016), reducing the density of trap states (Lu et al., 2017b) and preventing the undesirable chemical interactions (Kulkarni et al., 2016) using this strategy.

The wide band gap semiconductors, SnO₂ and WO₃ can be good candidates to be used as UTLs for optimizing TiO₂-based planar PSCs. Compared to TiO₂, several advantages have been reported for SnO₂ such as faster electron extraction, better energy band alignment to perovskite and more efficient electron transfer ability as well as minimizing/eliminating degradation of PSCs (Christians et al., 2018). WO₃ is also a chemically stable semiconductor and compared to TiO₂, indicates higher electron mobility (10–20 cm²·V⁻¹·s⁻¹), better transparency and a faster kinetic process of charge transfer at its interface with perovskite layer (Wang et al., 2015).

Accordingly, the main idea of this paper is to develop TiO₂/SnO₂ and TiO₂/WO₃ bilayers as ETLs by the precise, reproducible and industry compatible method of RF-sputtering and investigate how combining the merits of these oxide films affect the photovoltaic performance of TiO₂-based planar PSCs. This study can be of benefit for both efficiency improvement and further realization of the electron transport mechanism in planar PSCs. To further confirm the photovoltaic behaviors, the J-V results were compared using both spin-coated (SC) and RF-sputtered (RF) TiO₂ compact layers (CLs). Furthermore, the UTLs were examined with different thicknesses of 4 and 10 nm. A simulation work was also performed on the most efficient cells to obtain a deeper insight to device electrical properties. Finally, it was found that with these interfacial bilayers, photovoltaic characteristics and cell performance were improved noticeably.

2. Materials and experimental procedures

2.1. Cell preparation

2.1.1. Preparation of glass substrates

F:SnO₂ (FTO) substrates (Solaronix TCO22-15) were chemically etched with zinc powder and 2 M HCl solution and then cleaned through washing the substrates in several stages of sonication in soap solution, 0.2 M HCl solution in ethanol, acetone, ethanol and isopropanol for 10 min respectively and then dried in an oven of 120 °C for 30 min. All substrates were further cleaned by heating at 500 °C for 30 min.

2.1.2. Deposition of TiO₂, SnO₂ and WO₃ layers

RF TiO₂ CLs were deposited using Ti target (99.5%, 2 in. in diameter, 5 mm thick) (Loterios, Italy). Deposition by RF-sputtering method was conducted after 10 min pre-sputtering in a DST3-T magnetron sputtering instrument (Nano-Structured Coatings Co.). The atmosphere in the chamber was made of Ar as inert gas and a partial pressure of O₂ as reactive gas where their flows were set by MFC at 4 and 2 sccm, respectively. The total working pressure was 1.05 × 10⁻² Torr in the sputtering chamber. The sputtering power was 100 W, the substrate temperature was 200 °C and the distance between substrate and target was 7 cm.

In case of depositing SnO₂ layers, the parameters of RF-sputtering deposition method were same as TiO₂ CLs except for applying a ceramic target of SnO₂ (99.99%, 2 in. in diameter, 3 mm thick, Sindhauer Materials GmbH, Germany).

WO₃ layers were also deposited by RF-sputtering method similarly

with these differences that a W target (99.95%, 2 in. in diameter, 0.5 mm thick, Plansee, Germany) was used, the Ar and O₂ flows were equally kept 4 sccm during deposition and the total pressure in the chamber became 1.18 × 10⁻² Torr in this case.

In this work, TiO₂-CL, TiO₂-CL/SnO₂-UTL and TiO₂-CL/WO₃-UTL were the different studied ETL architectures (Fig. 1(a)). After depositing each of the oxide layers, a 30 min heating process at 500 °C was conducted. There are reports introducing 500 °C as the optimum temperature for annealing the ETLs of SnO₂ (Qiu et al., 2018), TiO₂ (Huang et al., 2017; Lu et al., 2017a) and WO₃ (Eze et al., 2017). As another effective factor, controlling the layers thicknesses play an important role in device performance (Mohammadian-Sarcheshmeh and Mazloum-Ardakani, 2018). We tried to obtain 50 nm TiO₂ CLs, since it has been found as an optimized size for this electron transport compact layer according to literature (Noh et al., 2018; Xu et al., 2015). In bilayer cases, the best performances have been achieved with UTLs of >10 nm thick (Xu et al., 2015; Yang et al., 2016b). Therefore, the interfacial effects of 4 and 10 nm UTLs were investigated here. The thickness was controlled by adjusting time with keeping all other sputtering parameters fixed during a layer deposition process. Using cross-sectional FE-SEM images of several samples, the deposition rate was measured to be ~3 Å/s, 1 Å/s and 0.13 Å/s for SnO₂, WO₃ and TiO₂, respectively. The deposition rates were confirmed by cross sectional FE-SEM images (Fig. S1) and found to be independent of substrate.

2.1.3. Preparation of TiO₂ by spin coating (SC)

SC TiO₂ CLs were prepared by spin-coating method (2000 rpm) on the pre-treated FTO substrates with 30 s deposition time. In this deposition stage, a 0.15 M Titanium iso-propoxide (TIP-Merck 97%) solution in anhydrous ethanol was normally used. Then a heating process at 500 °C for 30 min was carried out to prepare a severely attached TiO₂ CL on the FTO surfaces.

2.1.4. Cs/MA/FA perovskite precursor solution and film deposition

The perovskite layer with Cs_{0.05}(MA_{0.17}FA_{0.83})_{0.95}Pb(I_{0.83}Br_{0.17})₃ formula was deposited on the ETLs via one step deposition method. The precursor solution was prepared by adding FAI (CH₃NH₂)₂I (1 mmol) (Dyesol 98%), PbI₂ (1.5 mmol) (TCI 99.99%), MABr (CH₃NH₃Br) (1 mmol) (Dyesol 98%), and PbBr₂ (1.5 mmol) (IRASOL 99.99%) to N,N-dimethylformamide (DMF-Merck 99.5%) and dimethyl sulfoxide, (DMSO-Merck 99.9%) mixed anhydrous solvent (DMF:DMSO, 4:1 vol ratio). Then 50 μl of CsI solution containing CsI (1.5 mmol) (Sigma-Aldrich 99.999%) in DMSO (1 ml) was dissolved in 950 μl of aforementioned solution. This triple cation perovskite solution was then spin-coated on FTO/ETL substrates at 1000 rpm/10 s and 6000 rpm/30 s. In the last 15 s, 200 μl anti-solvent Ethyl acetate solution was poured slowly and layers were then quickly heated at 100 °C for 50 min.

2.1.5. Deposition of hole transport layer and top electrode

The Spiro-OMeTAD hole transport layer (HTL) was over-deposited by a 5000 rpm/30 s spin-coating process. The precursor solution was prepared by adding 72.3 mg of Spiro-OMeTAD (99.0%) in 1 ml of chlorobenzene (Merck 99%), 17.5 μl of a 520 mg mL⁻¹ solution of lithium bis(trifluoromethylsulfonyl)imide (Li-TFSI) (Sigma-Aldrich 99.5%) in acetonitrile and 28.8 μl of 4-*tert*-butylpyridine (TBP, Sigma-Aldrich 96%). Finally, 80 nm Au metallic contact was deposited in a high vacuum chamber using a DTT thermal evaporation instrument (Nano-Structured Coatings Co.).

2.2. Characterizations

The X-ray diffraction patterns were measured using an ASENWARE AW-DX300 equipment with Cu K_α (λ = 1.542 Å) radiation. Scanning electron microscopy (SEM) images were observed applying a TESCAN-Mira3 fieldemission scanning electron microscope (FE-SEM). UV – vis absorption spectroscopy was performed using a Perkin Elmer

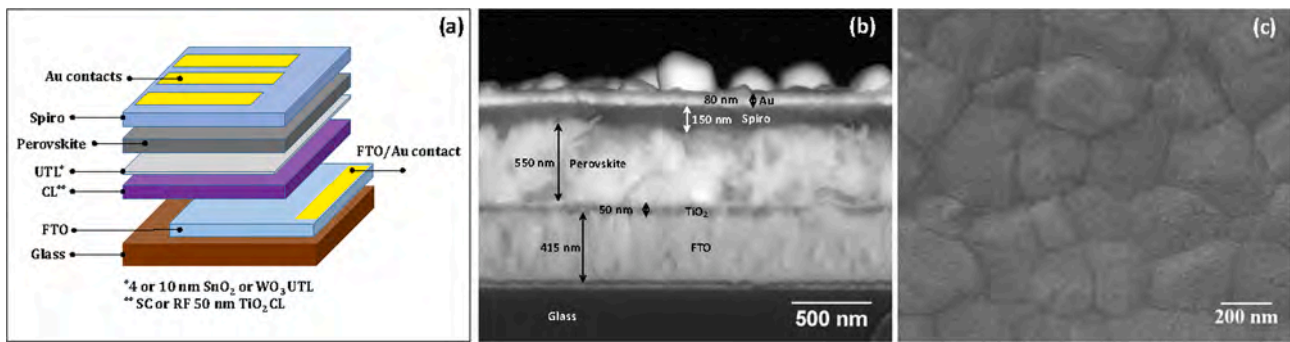


Fig. 1. (a) Schematic diagram of the device structure, (b) FE-SEM image of the cross section of SC TiO₂ based cell; (c) top view FE-SEM image of the perovskite layer.

Lambda 25 spectrophotometer. Atomic force microscope (AFM) measurements were obtained with ENTEGRA AFMNT – MDT and analyzed by Nova 1.0.26.1441 Pro. The Mott-Schottky analysis was taken in a quartz cell filled with 0.5 M Na₂SO₄ solution (pH 6.0) applying a Metrohm-Autolab potentiostat system with three-electrode configuration. The working, counter and reference electrodes were the fabricated metal oxide/FTO, Pt and Ag/AgCl (3 mol/kg KCl), respectively. The deposited metal oxide electrode included SnO₂, WO₃, RF TiO₂, or SC TiO₂ films with 50 nm thickness coated on FTO substrates for which the exposed areas were 0.65, 0.56, 0.78 and 0.75 cm², respectively. The frequency of the signal was 1000 Hz and scanning was done at dark with speed of 50 mV/s. All the above mentioned characterizations were performed on the layers after a 500 °C/30 min annealing process.

Measurements of incident photon-to-current conversion efficiency (IPCE) were performed utilizing a Sharif Solar IPCE-015 equipment. The photocurrent-voltage analyses were carried out by a Sharif Solar SIM-1000 system under AM 1.5 (100 mW/cm²) (1 sun) simulated light radiation (calibrated by a Thorlabs photodiode). The J–V curves were recorded with the scan rate of 5 mV s⁻¹ by a Keithley 2400 source meter. A mask with an open area of 0.09 cm² was used for masking the active area of the cells in these measurements. Electrochemical impedance spectroscopy (EIS) spectra were recorded using an Ivium, Compactstat potentiostat/galvanostat equipment.

3. Results

3.1. Structural, optical and electrical properties of ETLs

The surface morphology of TiO₂ CL/SnO₂ (4 nm) and TiO₂ CL/WO₃ (10 nm) bilayers on FTO by FE-SEM and AFM before and after modification by UTLs are shown in Fig. 2(a–d). Both the UTLs and TiO₂ CLs in the bilayer structures were deposited by RF-sputtering. A comparison of Fig. 2(a) with (b,c) reveals that the modified surfaces seem to have more uniform and sticking grains. Therefore, reduced grain boundaries and improved crystallinity, hence better conductivity (please refer to R_s values measured by EIS analysis), are expected for the modified CLs, facilitating the charge transport (Bu et al., 2017; Wang et al., 2017; Xu et al., 2015). The RMS surface roughness was measured to be 13.2 nm before modification, which reduced to about 10.5 nm after modification by both kinds of UTLs. The surface uniformity also improves the perovskite to ETL adhesion, and as a result, provides a better charge extraction and transport (Choi et al., 2016; Huang et al., 2017). From Fig. 2(a) and (d), clearer hence better crystallinity of the surface grains is observed for the RF TiO₂ CL compared to the SC layer. However, the spin coated TiO₂ CL Fig. 2(d) exhibits more uniform cloudy coverage with less RMS roughness of 12.0 nm. The smaller RMS roughness can be due to rotating of substrate during spin-coating, while this option was not available in the sputtering method.

The presence of the different UTLs was more investigated by energy dispersive X-ray spectroscopy (EDX) (Figs. S2(a–d)). From the figure and the inset tables, the addition of WO₃ or SnO₂ has led to the appearance of

W peaks or the Sn atomic ratio enhancement, respectively. The elemental proportion of O has been increased as well. The chemical composition of these layers could not be determined by EDX analysis as the beneath layers contributes to the signals.

The XRD characterization results of the fabricated layers are given in Fig. S3 and Table S1. Fig. S3(a) shows diffraction patterns of 50 nm SnO₂, WO₃ and TiO₂ films without sign of other oxide impurities. Fig. S3 (b) shows the XRD pattern of the fabricated perovskite layer for which the Cs_{0.05}(MA_{0.17}FA_{0.83})_{0.95}Pb(I_{0.83}Br_{0.17})₃ formula with perovskite structure is proved comparing with other reports (Saliba et al., 2016). This characterization along with the cross sectional FE-SEM image of the fabricated PSC (Fig. 1(b)) and the top view image of the coarse grained perovskite layer (Fig. 1(c)), demonstrates the formation of high quality perovskite layer (Gedamu et al., 2018).

Fig. 3(a) shows the UV–Vis transmittance spectra of the different deposited TiO₂ based ETLs on FTO. From the spectra, the transmittances of the discussed ETLs all exceed 80% in the whole visible region, which favors the sufficient absorption of light by the perovskite (Wang et al., 2015). Therefore, the treatment of the TiO₂ layer does not affect the transmittance due to the very small thickness of WO₃ or SnO₂. Moreover, the peak and valley in the spectra reveals a good layer smoothness and uniformity (Swanepoel, 1984). It is known that the ETL band gap is an important factor seriously influences the cell performance. Therefore, using the UV–Vis data, the band gap of the fabricated 50 nm CLs was calculated based on the Tauc equation (Fig. 3(b)):

$$\alpha h\nu = B(h\nu - E_g)^m \quad (1)$$

In Eq. (1), $m = 2$ for allowed indirect and $m = 1/2$ for allowed direct transitions. For WO₃ and TiO₂, the best linear fitting was achieved taking $m = 2$, which is in agreement with other reports for 500 °C annealed WO₃ (Ganbavle et al., 2014) and TiO₂ (Lim et al., 2014; Reyes-Coronado et al., 2008). The calculated band gaps of the layers are presented in Table 1. The band gap of SnO₂ could not be calculated as it shows no absorption shoulder.

Fig. 4 shows the Mott-Schottky plots of the 50 nm TiO₂, SnO₂ and WO₃ thin films on FTO. The positive slope of the curves indicates that they are n-type semiconductors. The N_D and V_{fb} is calculated using the Mott-Schottky equation:

$$\frac{1}{C^2} = \frac{2}{\epsilon_0 \epsilon_r e N_D} \left(V - V_{fb} - \frac{kT}{e} \right) \quad (2)$$

in which e , ϵ_r , ϵ_0 , V_{fb} , V and N_D are electron charge, dielectric constant of semiconductor, vacuum permittivity, flat band potential, applied potential and carrier concentration, respectively (Li and Wu, 2015). It is also possible to specify the position of the conduction band minimum based on the Eq. (3), using the N_D and the flat band values:

$$E_C = V_{fb} + kT \ln \left(\frac{N_D}{N_C} \right) \quad (3)$$

where N_C is the effective density of states in conduction band. Then the

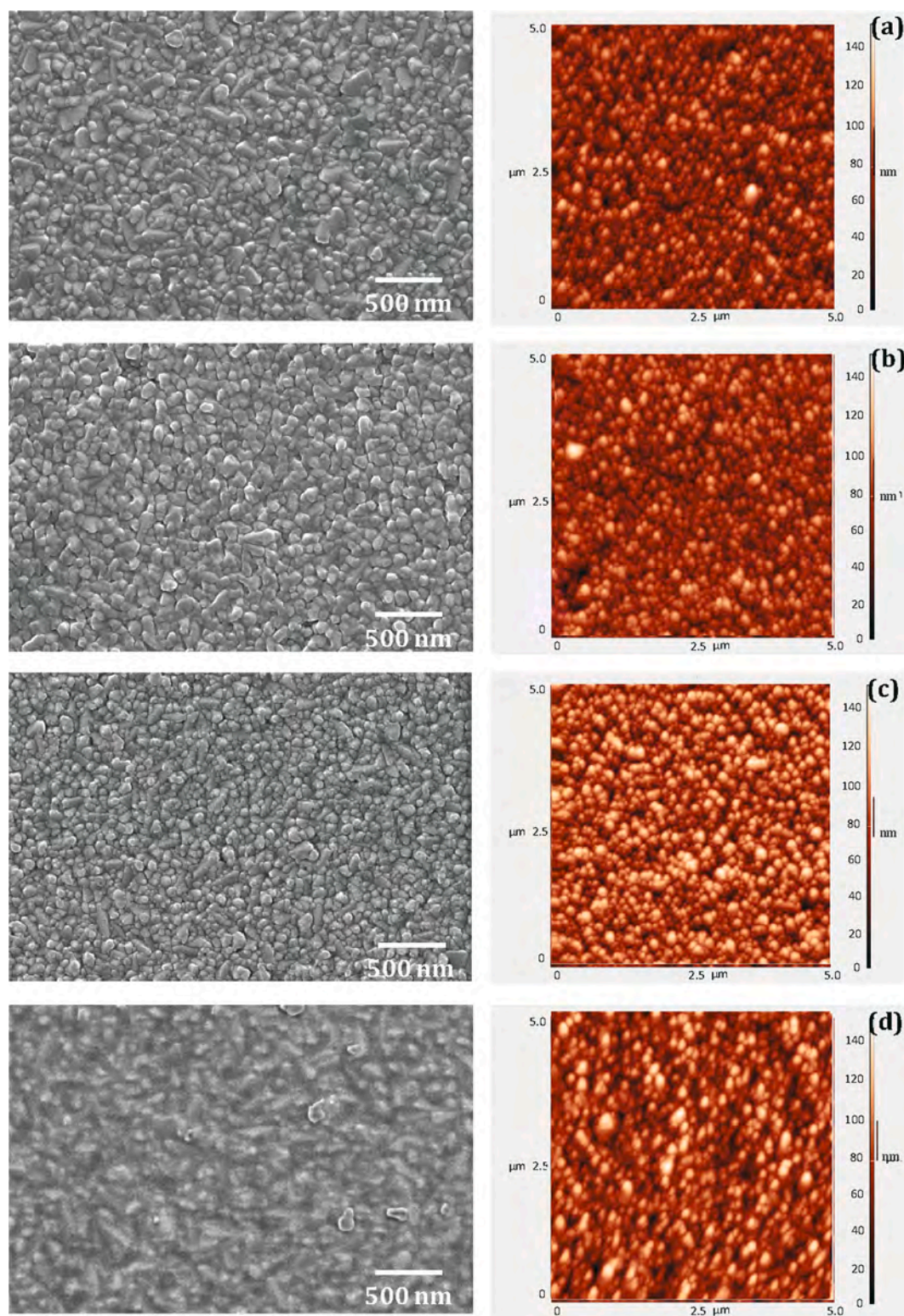


Fig. 2. Top view FE-SEM and AFM (5 μm \times 5 μm) images of (a) the bare RF TiO₂ CL film, (b) TiO₂ CL/SnO₂ UTL and (c) TiO₂ CL/WO₃ UTL bilayers as well as (d) the bare SC TiO₂ film.

valance band maximum is then achieved by inserting the calculated E_C and band gap into $E_V = E_C - E_g$ equation (Haghighi et al., 2018). The energy band edges calculated based on the Mott-Schottky and UV-Vis results are presented in Table 1, which is used to sketch the band alignments. In addition, the measured N_{DS} and V_{fb} as well as the input values of ϵ_r and N_c for Mott-Schottky calculations (Eqs. (2), (3)) can be

found in Table S2.

The electron transport ability of the modified and unmodified TiO₂ CL was also comparatively evaluated by the photoluminescence (PL) measurements (Fig. S4). It was found that the bilayer structured ETLs exhibit an obvious PL quenching, suggesting higher electron extraction ability, due to the conduction band edge upshift and also enhanced

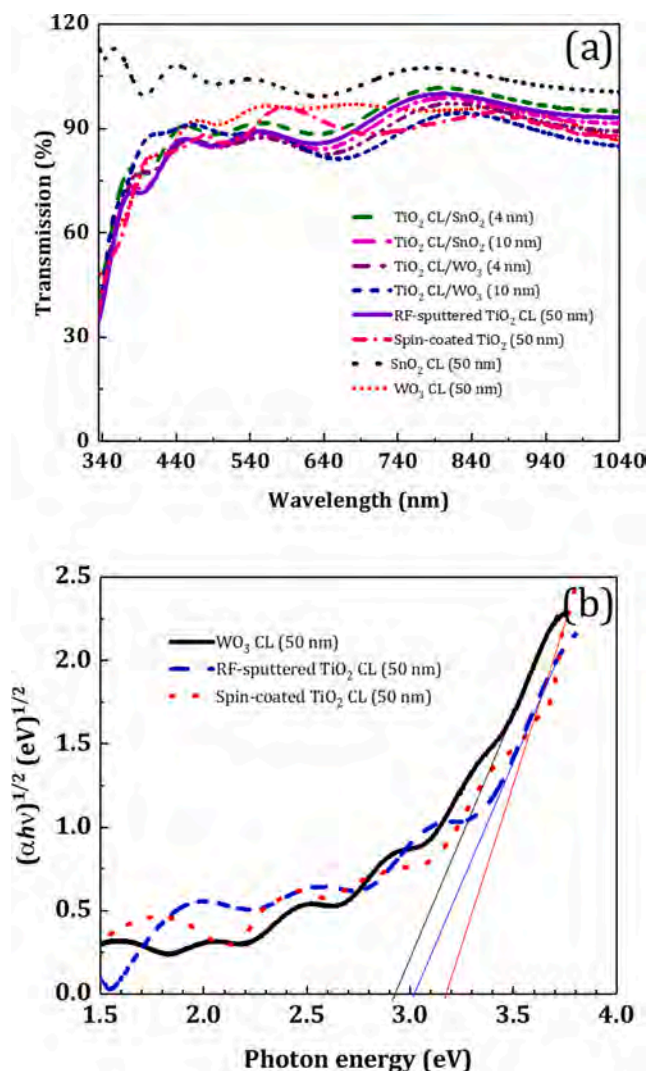


Fig. 3. (a) UV-Vis spectra of the single layers and RF TiO₂ based bilayers, (b) Tauc plots of the fabricated metal oxide semiconductors.

Table 1

The measured energy band edges of the fabricated metal oxide semiconductors.

E _v (eV)	E _g (eV)	E _c (eV)	Layer
-8.09	3.20	-4.89	SC TiO ₂
-7.95	3.01	-4.94	RF TiO ₂
-8.19	3.50*	-4.69	SnO ₂
-7.51	2.92	-4.59	WO ₃

* Based on Ref. (Qiu et al., 2018).

interface area between perovskite and ETL. Interestingly, a further degree of PL quenching has been achieved with SnO₂ UTL, indicating the most impactful electron picking up as a result of the more efficient energy band alignment.

3.2. Photovoltaic properties of the devices

Fig. 5(a,b) show J-V curves for the champion cells with RF SnO₂ and WO₃ UTLs (4 and 10 nm) on 50 nm SC or RF TiO₂ CLs, respectively. Table 2 depicts the corresponding calculated device parameters. The statistics of the measured devices is also revealed in Fig. S5. Since the backward (BW) scans provide efficiencies higher than the forward (FW) scans, the following focuses mainly on the reverse scan parameters. From the figure and table, embedding of the UTLs affects the J-V

characteristics of the cell in a same manner for both SC and RF TiO₂ CLs. One point is that PCE increases by this modification and the thinner UTLs of 4 nm have greater effect on the efficiency compared to 10 nm UTLs. In addition, the SnO₂ UTLs improve the PCE further than the WO₃ UTLs, enhance the J_{SC} while keep the V_{OC} almost unaffected. Besides, one can see that although the WO₃ UTLs reduce the V_{OC}, they significantly enhance the J_{SC} so that the efficiency overall feature improves. As these findings could be observed for each TiO₂ deposition techniques, the effective role of our UTLs on J-V parameters is better confirmed.

From Table 2 it can be observed that SC TiO₂ layers exhibit better PCE values. However, independent of the deposition method, the relatively low PCE in bare TiO₂ cells is mainly due to the lower J_{SC} values, regarding that V_{OC}s of the UTL containing cells are not greater. Moreover, a serious hysteresis accompanied by a strange bending is also observed in J-V curve of the bare SC TiO₂ cell in the 0–600 mV range. This kind of bending/hysteresis has been also reported elsewhere for the PSCs made by bare SC TiO₂ CLs, attributed to the slow extractions of charges and their accumulation at the interfaces (Lu et al., 2017b; Xu et al., 2015). Accordingly, our data with the existing reports suggest it is likely that the unmodified bare TiO₂ CL is not suitable for planar PSCs.

Our J-V results also represent the impact of the deposition layer method and UTL modification material on the hysteresis behavior. Table S3 depicts the calculated hysteresis index of the cells corresponding to Fig. 5. It can be seen that the RF TiO₂ based cells show a significantly lower hysteresis, which could be due to the better crystallinity of the CLs deposited by the sputtering technique (see Fig. 2) (Chen et al., 2016). Furthermore, the WO₃-UTL cells show an overall less hysteresis effect, especially for 10 nm thickness, compared to the SnO₂-UTL cells, which can be attributed to the better quality of the perovskite absorber layer according to literature (Gheno et al., 2017).

The J-V results reveal special photovoltaic behaviors for WO₃ PSCs. It can be seen that although WO₃ UTLs reduce the V_{OC}, it considerably increases the J_{SC}, which eventually leads to cell's efficiency enhancement. This effective role of WO₃ UTL on the efficiency and J_{SC} improvement was tested and confirmed again by comparison of IPCE measurement for the two cells with and without WO₃ UTL (Fig. S6). By integration over the whole spectra in this figure, we calculated J_{SC} to be 19.61 and 8.73 mA/cm², which confirms almost the data reported in Table 2. These distinctive J-V characteristics are similarly observed for the planar PSCs containing the bare WO₃ CL compared to those including TiO₂ or SnO₂ CL (Fig. S7). The WO₃ PSCs also remarkably reveal a further degree of degradation (Fig. S8(a,b)). From the figure, the WO₃-UTL cell shows 34% efficiency decay after two weeks, while it is 21% and 10% for the SnO₂-UTL and bare TiO₂-CL cells, respectively. The dark J-V curves of the RF TiO₂ based PSCs has been also shown in Fig. S9(a). It can be seen that a kind of step is observed in the dark J-V curves of the cells with a WO₃ UTL. For more investigation, the dark J-V curves of some other comparative cells fabricated and studied in our research has been also indicated in Fig. S9(b). This step is also observed in other dark J-V curves of the cells containing a WO₃ thin or ultra-thin layer while the dark J-V curves of those with SnO₂ or/and TiO₂ layer do not show such a behavior.

To further study the role of the inserted UTLs in the performance of planar PSCs, electrochemical impedance spectroscopy (EIS) curves were recorded for the most efficient cells (the SC TiO₂ based champion cells), at V_{OC} under 1 sun illumination (Guerrero et al., 2016). This result is useful for tracking the transport and recombination dynamics of internal carriers and for quantifying the parameters of charge transport, such as chemical capacitance, recombination resistance, and charge conductivity in planar PSCs (Chandiran et al., 2014). Fig. 6 indicates the Nyquist diagrams of the PSCs with the studied ETL types based on TiO₂ CL. It can be seen that all the diagrams consist of one pronounced arc in low frequency region in addition to a small arc at high frequencies. A simulated circuit model, obtained by the Zview software, fits the Nyquist plots (inside Fig. 6). The high-frequency small arc (left side) radius is directly proportional to the charge transport resistance (R_{CT}) at the

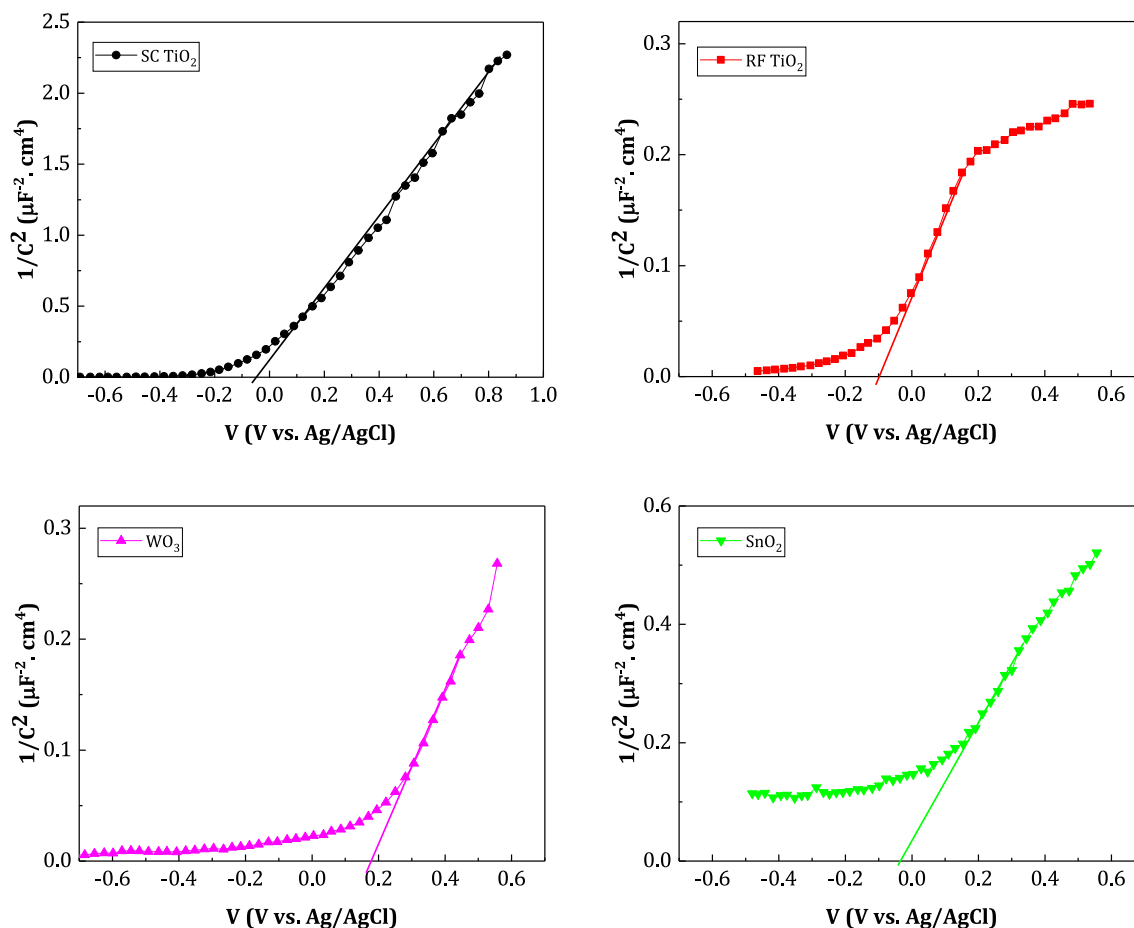


Fig. 4. Mott-Schottky analysis of the fabricated metal oxide semiconductors with 50 nm thickness coated on FTO substrates.

perovskite/ETL interface (Huang et al., 2017). While the low frequency arc (right side) is a transmission line with radius proportional to recombination resistance, R_{rec} , (inversely proportional to the recombination rate) at the perovskite/ETL interface (Huang et al., 2017). Table 3 depicts the calculated fitted values. From the inset circuits, a CPE (constant phase element) is used instead of C (capacitor) to model the depressed recombination semi-circles for the modified cells. This is attributed to the bilayer structures causing a double-layer capacitance, which appears as a CPE (Barsoukov and Macdonald, 2005). Since the cells differ only in the UTLs, any change in the EIS parameters, specially R_{CT} and R_{rec} , can be related to the difference in the ETL/perovskite interface. This is more distinctively observed in the corresponding capacitance-frequency diagram (Fig. S10) in which the plots overlap in high frequency range and separate in low frequency range (Guerrero et al., 2016).

From the inset of Fig. 6 and Table 3, it can be claimed that PSCs containing TiO_2/SnO_2 bilayers have lower charge transfer resistance, R_{CT} , compared to UTL free bare TiO_2 cell which could be partially due to an increased adhesion between the perovskite and composite layers (see the AFM images). This allows for more efficient extraction of the photogenerated electrons (Lu et al., 2017b). On the other hand, a larger energy gap between TiO_2 and the perovskite absorber in the bare TiO_2 -based creates a larger energy barrier (Fig. 7) for electron transfer hence a larger R_{CT} (Huang et al., 2017). Inserting a the UTLs upshift the band edge and reduce the R_{CT} (Huang et al., 2017). It can be also noted that increasing the thickness of UTLs from 4 to 10 nm leads to larger R_{CT} s as it prolongs the perovskite to TiO_2 layer charge diffusion length, declines the carrier transfer and increases the interface charge accumulation (Lu et al., 2017b).

Based on the right semicircles, the effect of the UTLs on the

recombination resistance of the cells is determined. Comparison of the EIS curves shows that bare TiO_2 cell has a strong charge recombination, due to smaller radius hence lower R_{rec} (Table 3). It can be clearly seen that R_{rec} increases in the presence of UTLs, meaning that the recombination rate has been reduced. It is likely that in the absence of UTLs, randomly diffused carriers gather and recombine at the ETL/perovskite interface and, as a result, R_{rec} decreases (Huang et al., 2017). When a UTL is introduced between perovskite and ETL, the photogenerated charges transfer more effectively and then the recombination is sharply suppressed. As a result, in case of TiO_2/SnO_2 ETLs, the reduced R_{CT} and increased R_{rec} imply more efficient extraction/transport of the electrons leading to increased J_{SC} and PCE for these bilayer structured planar PSCs.

Another interesting point is that value of the starting point at the real part of the Nyquist plot corresponds to the series resistance R_s (Guerrero et al., 2016). The value of R_s is determined by the resistance of the FTO substrate, external wires, and other additional contributions. Other reports demonstrate that the ETL also have an effect on R_s (Gonzalez-Pedro et al., 2014). From Fig. 6, it is observed that the R_s of all TiO_2 based devices are approximately the same as well as expected since the presence of a UTL with a low thickness introduces a negligible resistance. The little reduction in R_s values (Table 3) can be attributed to the improved adhesion between perovskite and the bilayers.

3.3. Efficiency enhancement mechanism

The efficiency enhancement achieved by inserting the UTLs of SnO_2 or WO_3 can be interpreted by two different strategies. One reason for the improvement of J_{SC} and subsequently PCE as well as electron extraction is considered owing to the improved adhesion between perovskite layer

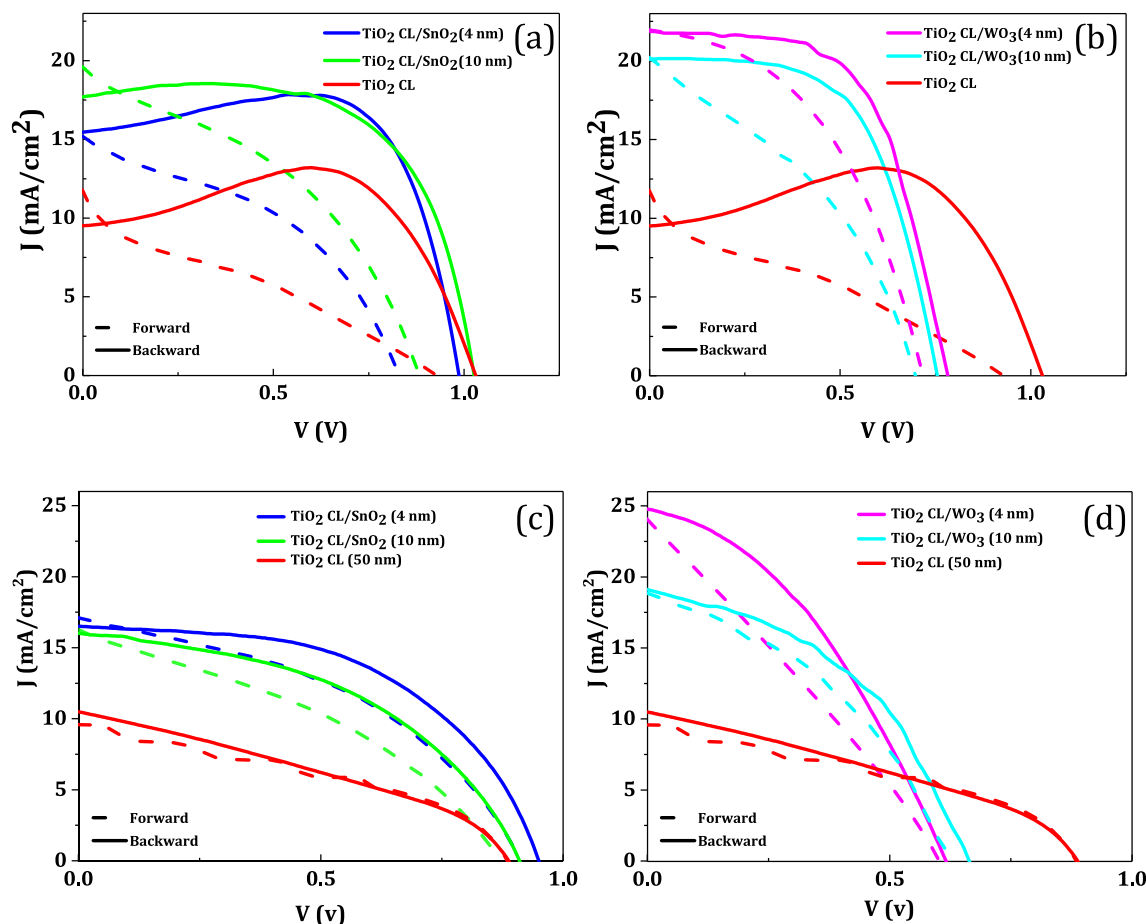


Fig. 5. J–V characteristics of the champion devices based on (a,b) SC TiO₂ and (c,d) RF TiO₂ with (a,c) SnO₂ UTL and (b,d) WO₃ UTL, under 1 sun illumination.

Table 2
Photovoltaic parameters of the TiO₂ based champion PSCs.

UTL	TiO ₂ CL	Scan direction	J _{sc} (mA/cm ²)	V _{oc} (V)	FF	PCE (%)
SnO ₂ (4 nm)	SC	BW	15.45	0.99	0.80	12.30
		FW	21.03	0.83	0.30	5.25
	RF	BW	16.74	0.95	0.52	8.23
		FW	17.11	1.00	0.39	6.65
SnO ₂ (10 nm)	SC	BW	17.68	1.03	0.66	11.96
		FW	19.61	1.10	0.32	6.97
	RF	BW	16.42	0.91	0.45	6.71
		FW	16.30	1.00	0.32	5.23
WO ₃ (4 nm)	SC	BW	21.88	0.78	0.60	10.24
		FW	22.16	0.71	0.46	7.26
	RF	BW	25.21	0.62	0.38	5.82
		FW	24.08	1.00	0.17	4.01
WO ₃ (10 nm)	SC	BW	20.10	0.75	0.60	9.10
		FW	21.96	1.10	0.30	5.30
	RF	BW	19.98	0.66	0.41	5.50
		FW	18.87	0.62	0.39	4.64
no UTL	SC	BW	10.45	1.03	0.75	8.18
		FW	11.55	0.88	0.33	3.42
	RF	BW	10.48	0.89	0.34	3.16
		FW	9.58	0.89	0.37	3.12

and the bilayer with less interfacial trap states, facilitating charge extraction and reducing carrier recombination (Huang et al., 2017). The significant differences in film morphology as discussed in Fig. 2 (AFM and SEM) can be one of the origins. On the other hand, more effective band alignment of cells with UTLs of SnO₂ or WO₃ is likely the main root

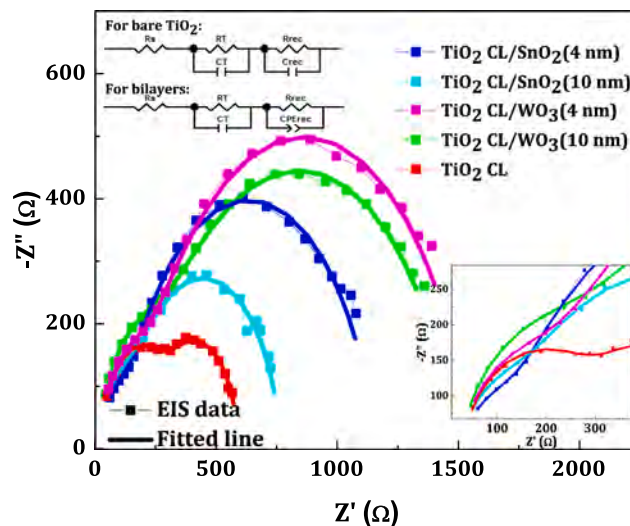


Fig. 6. Nyquist plot of the PSCs using different TiO₂ based ETLs. The equivalent circuits of this study as well as the zoomed feature of the high frequency semicircles are shown in the inset. CPE denotes constant phase element, and C denotes capacity.

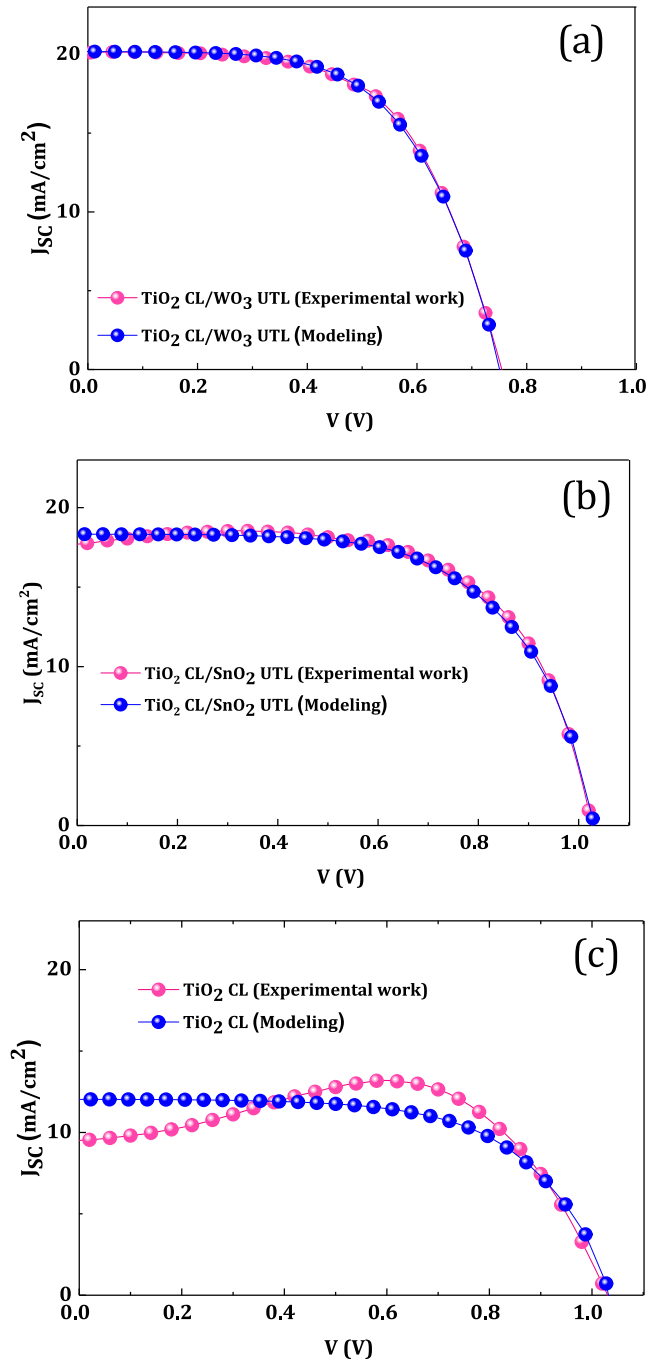
of the improved electron transport.

Scheme 1(a,b) shows the schematics of energy band alignment in the TiO₂-based bilayer structured PSCs based on the UV–Vis and Mott–Schottky results (Table 1). It reveals that both the VBM (valence band maximum) and CBM (conduction band minimum) of WO₃ (Scheme 1(a))

Table 3

Fitted resistance values calculated based on EIS measurements.

Resistance	SnO ₂ (4 nm)	SnO ₂ (10 nm)	WO ₃ (4 nm)	WO ₃ (10 nm)	no UTL
R _{CT} (Ω)	98	134	166	207	270
R _{rec} (Ω)	1043	614	1373	1274	290
R _S	19	19	18	16	28

**Fig. 7.** The experimental and simulated diagrams of the TiO₂ based cells with (a) WO₃ UTL, (b) SnO₂ UTL and (c) no UTL.

are a little higher than those of TiO₂, suggesting that the TiO₂/WO₃ bilayer has a type-II energy band structure (Lu et al., 2017b; Xu et al., 2015). This helps the photo-induced carriers to transport without facing an energy barrier (Huang et al., 2017) hence avoiding charge

accumulation and as a result recombination reduces (Roh et al., 2006).

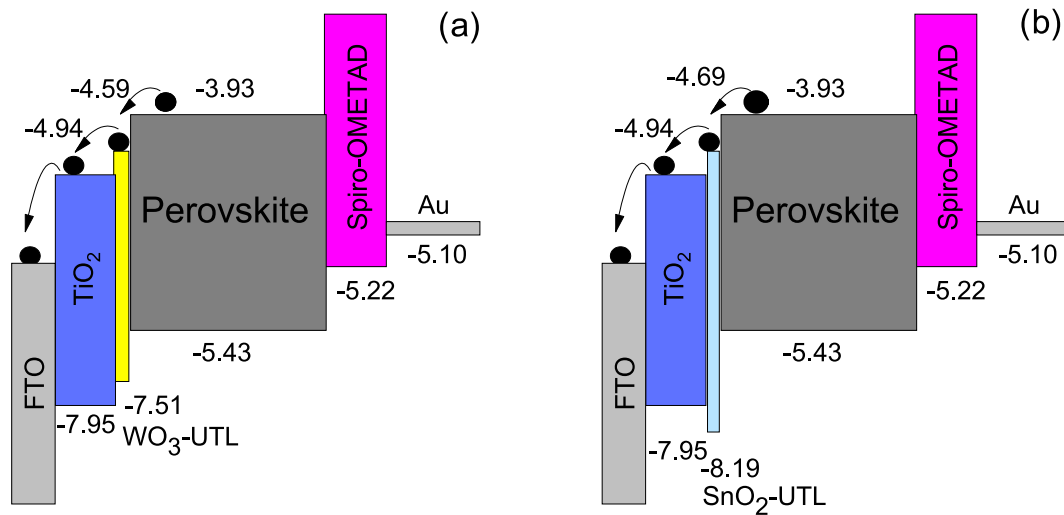
In case of the SnO₂-UTL PSCs (Scheme 1(b)), the band alignment is of benefit in two respects. First, like WO₃ UTLs, the generated electrons are more easily injected because of the higher CBM edge. Second, it blocks the transfer of holes from perovskite to TiO₂ CL owing to a lower VBM (Kulkarni et al., 2016), leading to less charge recombination. Therefore, higher efficiencies obtained using the UTLs of SnO₂ compared to that of WO₃ (Fig. 5(a,b)) is related to the more effective band alignment leading to a remarkably decreased charge recombination.

3.4. Modeling with SCAPS-1D

To further study the contribution of WO₃ or SnO₂ UTLs on the electrical properties of the perovskite layer and the interfaces, the most efficient cells (the SC TiO₂ based champion cells) with the three ETL types of bare TiO₂ CL (50 nm), TiO₂ CL/SnO₂ UTL (10 nm) and TiO₂ CL/WO₃ UTL (10 nm) were modeled using Solar Cell Capacitance Simulator (SCAPS) and compared with experimental results (Fig. 7(a–c)). Due to the quantum size effects (for thicknesses less than 5 nm) (Chouchen et al., 2019; Hossain et al., 2018), the devices with 4 nm UTLs were not used in this simulation. A 1.5 AM solar radiation with a power density of 100 mW/cm² were used as the source of illumination and temperature was set as 300 K. In this work, an internal defect near the midgap states has been assumed for each layer. The light reflection of the front and the back contacts were set to be 0.05 and 1, respectively. The input values listed in Table 4 were used to calculate the performance of the cells. No radiative band-to-band recombination and Auger recombination are considered. The only recombination mechanism considered here is defect induced (SRH-Model). Tables S4 to S6 list the values of defect properties which are mostly obtained based on fitting.

The fittings based on the simulation results reveal that the defect formation in the bulk of the perovskite depends on the cell ETL type. Table S4 shows the concentration and energy levels of the defects. However, independent of the cell type, they remain unchanged in other layers of the device (Table S5). For the WO₃-UTL cell, the energy level and the concentration of the perovskite single acceptor defects were obtained to be 0.05 eV and $7.0 \times 10^{14} \text{ cm}^{-3}$, respectively. According to the literature, these values suggest that the acceptor defect I_i is dominant (Han et al., 2017). In the case of donor defects, energy level and donor concentration were found to be 1.03 eV and $4.3 \times 10^{15} \text{ cm}^{-3}$, respectively which propose Pb_i is dominant. Similar quantities were obtained for the surface energy and concentration in SnO₂-UTL cell, indicating formation of similar dominant defects. In case of the bare TiO₂ device, the defect concentration is in order of 10^{10} cm^{-3} , which is negligible in comparison with UTL containing cells. There are probably unknown defects like neutral defects or clusters (Chen et al., 2013) that can affect electron affinity and optical band gap, leading to a lower absorption constant in the absorber layer (Table 4) (Niemegeers et al., 1998). Therefore, our results predict existence of new types of defects in the perovskite layer of bare TiO₂ based planar PSCs, which has been rarely noticed by scientists and reported.

The observed special photovoltaic properties of the WO₃-UTL PSCs, can be explained based on the simulation. Our modeling with WO₃ UTL shows that the defect concentration at perovskite/ETL interface is remarkably higher compared to the SnO₂-UTL or bare TiO₂ devices (Table S6). This is responsible for the lower V_{OC} and FF in the WO₃ PSCs, which has also been reported elsewhere (Eze et al., 2017; Wang et al., 2018). Our fitting results also suggest that the greater J_{SC} in WO₃ PSCs originates from a higher integral photogeneration rate (Fig. 8). This conclusion is also found in literature where the enhanced J_{SC} in the presence of WO₃ has been attributed to a fairly better behavior of the perovskite absorber layer due to its superior crystallization which leads to an increased photoabsorptions, hence an improved photogeneration rate (Wang et al., 2015).



Scheme 1. The band alignment schematics of PSCs with (a) TiO₂ CL/WO₃ UTL and (b) TiO₂ CL/SnO₂ UTL bilayers.

Table 4
Basic data for the modeling case.

Parameters and units	TiO ₂ CL	SnO ₂ UTL	WO ₃ UTL	Perovskite			HTL-Spiro
				TiO ₂ CL/WO ₃ UTL	TiO ₂ CL/SnO ₂ UTL	TiO ₂ CL	
Thickness (nm)	50 (exp)*	10 (exp)	10 (exp)	550 (exp)	550 (exp)	550 (exp)	150 (exp)
Electron affinity (eV)	4.89 (exp)	4.69 (exp)	4.59 (exp)	3.93 (Shen et al., 2017)	3.93 (Shen et al., 2017)	3.7 (fitting)**,(Shen et al., 2017)	2.2 (Adhikari et al., 2016; Ng et al., 2018)
Bandgap (eV)	3.2 (exp)	3.5 (Qiu et al., 2018)	2.92 (exp)	1.5 (Shen et al., 2017)	1.5 (Shen et al., 2017)	1.65 (fitting),(Shen et al., 2017)	2.91 (Adhikari et al., 2016; Ng et al., 2018)
Dielectric permittivity (relative)	31 (Berberich and Bell, 1940)	7.25 (Yildirim et al., 2014)	5.76 (Paliwal et al., 2014)	6.5 (Da et al., 2018)	6.5 (Da et al., 2018)	6.5 (Da et al., 2018)	3 (Shen et al., 2017)
CB effective density of states (N _C) (cm ⁻³)	6.67 × 10 ²⁰ (Kormann et al., 1988)	3.61 × 10 ¹⁸ (Button et al., 1971)	1.96 × 10 ¹⁹ (Berak and Sienko, 1970)	5.0 × 10 ¹⁸ (Shen et al., 2017)	5.0 × 10 ¹⁸ (Shen et al., 2017)	5.0 × 10 ¹⁸ (Shen et al., 2017)	1 × 10 ²⁰ (Shen et al., 2017)
VB effective density of states (N _V) (cm ⁻³)	6.67 × 10 ²⁰ (Kormann et al., 1988)	3.61 × 10 ¹⁸ (Button et al., 1971)	1.96 × 10 ¹⁹ (Berak and Sienko, 1970)	5.0 × 10 ¹⁸ (Shen et al., 2017)	5.0 × 10 ¹⁸ (Shen et al., 2017)	5.0 × 10 ¹⁸ (Shen et al., 2017)	1 × 10 ²⁰ (Shen et al., 2017)
Electron mobility (cm ² /Vs)	1.0 × 10 ⁻⁴ (Shen et al., 2017)	2.4 × 10 ² (Jiang et al., 2017; Van Zeghbroeck, 1997)	10 (Berak and Sienko, 1970)	2 (Shen et al., 2017)	2 (Shen et al., 2017)	2 (Shen et al., 2017)	2 (Shen et al., 2017)
Hole mobility (cm ² /Vs)	1.0 × 10 ⁻⁴ (Shen et al., 2017)	2.2 × 10 ² (Jiang et al., 2017; Van Zeghbroeck, 1997)	10 (Berak and Sienko, 1970)	2 (Shen et al., 2017)	2 (Shen et al., 2017)	2 (Shen et al., 2017)	1.0 × 10 ⁻² (Shen et al., 2017)
Shallow uniform donor density (N _D) (cm ⁻³)	2.81 × 10 ¹⁷ (exp)	2.42 × 10 ¹⁹ (exp)	3.68 × 10 ¹⁹ (exp)	0 (Shen et al., 2017)	0 (Shen et al., 2017)	0 (Shen et al., 2017)	0 (Shen et al., 2017)
Shallow uniform acceptor density (N _A) (cm ⁻³)	0 (exp)	0 (exp)	0 (exp)	0 (Shen et al., 2017)	0 (Shen et al., 2017)	0 (Shen et al., 2017)	1.0 × 10 ¹⁸ (Shen et al., 2017)
Absorption constant A (1/cm eV ^(1/2))	1.0 × 10 ⁴ (fitting)	1.0 × 10 ⁴ (fitting)	1.0 × 10 ⁴ (fitting)	4.0 × 10 ⁴ (fitting)	4.0 × 10 ⁴ (fitting)	3 × 10 ⁴ (fitting)	1.0 × 10 ⁴ (fitting)
Absorption constant B (eV ^(1/2) /cm)	0 (fitting)	0 (fitting)	0 (fitting)	0 (fitting)	0 (fitting)	0 (fitting)	0 (fitting)
Electron thermal velocity (cm/s)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)
Hole thermal velocity (cm/s)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)	1.0 × 10 ⁷ (Adhikari et al., 2016)

* From our experimental measurement.

** From our fitting process.

4. Conclusions

TiO₂/SnO₂ and TiO₂/WO₃ electron transport bilayers, which combine the merits of both SnO₂ or WO₃ and TiO₂ together have been

independently and successfully applied for the planar PSCs. From characterizations and simulation, the improved performance of the bilayer ETL based devices is attributed to the enhanced electron extraction, suppressed recombination rate and passivated defect states.

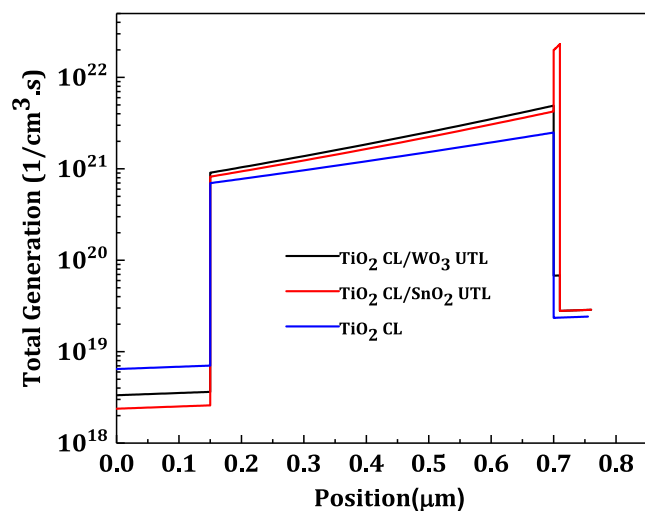


Fig. 8. A comparison between the photogeneration rates of the TiO₂ based cells.

These are consequences of improved adhesion between the perovskite and ETL, more effective band alignment and better formation of the absorber layer. To further confirming the results, the J-V curve changes were examined on similar bilayer structured PSCs based on TiO₂ CLs deposited by spin-coating and RF-sputtering, where same results were achieved. The best efficiency improvement was obtained for TiO₂-CL/SnO₂ (4 nm) bilayer (from 8.18% to 12.3%) owing to its well-aligned band structure for blocking holes. It was also revealed that thinner UTLs are more effective in PSCs with electron transport bilayer structures. The present work proved that performance of photovoltaic devices can be simply improved using the metal oxide interfacial bilayer structures.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Research data

The SCAPS simulation data codes are available in research data with SCAPS format and additional files that give access for everyone to simulate our work.

Appendix A. Supplementary material

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.solener.2020.08.035>.

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